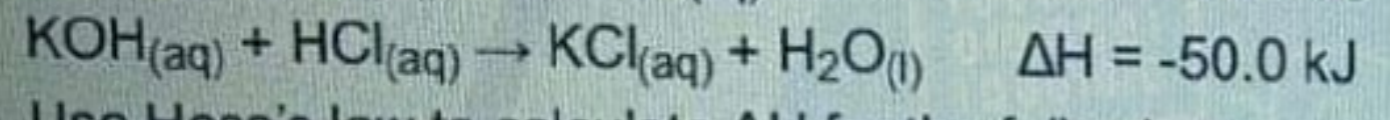
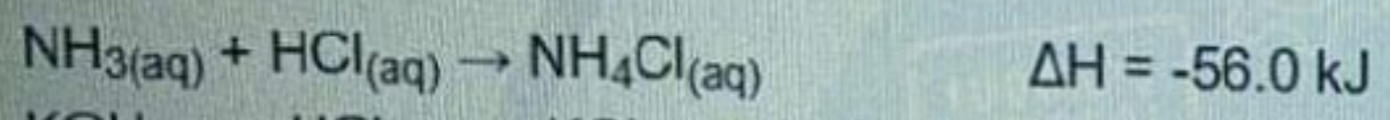
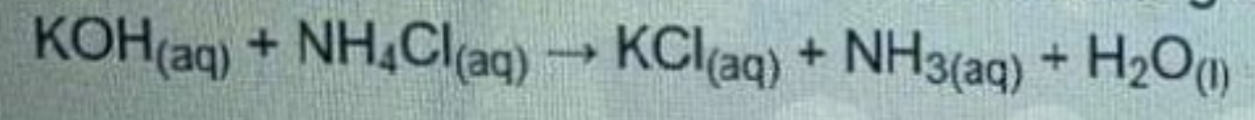


Given that:



Use Hess's law to calculate ΔH for the following reaction:



- +6.0 kJ
- +106.0 kJ
- 4.0 kJ
- 6.0 kJ
- 106.0 kJ

□

